

Capillary gas chromatographic separations of a multi-component mixture of polyalcohol compounds

Janusz Madaj, Andrzej Wiśniewski, Eugenia Skorupowa and Janusz Sokołowski*

Department of Chemistry, University of Gdańsk, Sobieskiego 18, 80-952 Gdańsk (Poland)

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ABSTRACT

Capillary gas chromatographic separation conditions were established for a 31-component mixture of polyhydroxy compounds containing two tetrityls, three pentitols, six hexitols, seven monoanhydropentitols, eight monoanhydrohexitols, three dianhydrohexitols, pentaerythritol and myoinositol. Mixtures of per-O-acetyl or per-O-trimethylsilyl derivatives of these compounds were separated on fused-silica columns using highly polar DB-23 and non-polar HP-5 stationary phases, respectively. Relative retention times for each compound with respect to pentaerythritol and myoinositol and methylene units were determined.

INTRODUCTION

Polyhydroxy compounds are widespread in the plant kingdom [1,2] and there is great interest in their detection and identification. High-resolution gas chromatography (HRGC) offers the possibility of the rapid detection of these compounds or their derivatives in mixtures from natural sources or resulting from chemical transformations of polyols.

The recent literature covering applications of GC to these compounds can be divided into three groups. The first is concerned with improvement of GC separation methods for alditol, anhydroalditol and cyclitol derivatives [3–19]. The second involves papers dealing with the detection and determination of these compounds occurring in nature, *e.g.*, in biological fluids [20,21], human cerebrospinal fluid [22], the blood serum [23], urine of patients with uraemia [24], human urine (6-deoxyallitol and 6-deoxyglucitol) [25], eyeballs of animals and healthy humans [26], eyeballs of patients with cataract

[27], animal cell plasma [28], cow milk [29], clover and peanuts [30], seeds and fodder (methylinositol) [31], raw food [32], foodstuffs [33] and plasticizers [34]. The third relates to the GC separation of enantiomers on columns with chiral stationary phases [35–39].

Our previous studies concerning the separation and identification of anhydroalditol mixtures formed in cyclization–dehydration reactions of particular pentitols [40–44] and hexitols [45,46] were performed using capillary GC (cGC) with SE-52, OV-275, Carbowax 20M and 20MTPA stationary phases and helium or hydrogen as the carrier gas. Based on these results, we were able to separate more complicated mixtures.

In this paper we present the results of the separation of per-O-acetyl and per-O-trimethylsilyl derivatives of a 31-component mixture of polyols by cGC.

EXPERIMENTAL

Gas chromatographic conditions

The 31-component mixture of polyhydroxy compounds was separated on two columns: as per-O-acetyl derivatives on a DB-23 fused-silica

* Corresponding author.

column (60 m × 0.258 mm I.D.) and as per-O-trimethylsilyl derivatives on an HP-5 fused-silica column (50 m × 0.2 mm I.D.).

Both columns were separately installed inside a Carlo Erba Vega 6180 gas chromatograph. The gas chromatograph was equipped with a cold on-column injector, a flame ionization detector and an integrator (CE Instruments DP 700). Hydrogen was used as the carrier gas at a flow-rate of 2 ml/min measured at room temperature. The temperature programme for the HP-5 column was from 115 to 180°C at 3°C/min and from

180 to 250°C at 4°C/min, and that for the DB-23 column was from 80 to 150°C at 3°C/min, from 150 to 200°C at 4°C/min and from 200 to 240°C at 5°C/min with a final hold at 240°C for 15 min.

Reactants

The following compounds were used: allitol, obtained by reduction of D-allose (Aldrich) with NaBH₄; D-altritol and galactitol, obtained by reduction of D-tagatose (Sigma) with NaBH₄; 1,4-anhydro-D-arabinitol [47]; 1,5-anhydro-D-arabinitol [47]; 1,4-anhydro-D,L-galactitol [47];

TABLE I
RETENTION CHARACTERISTICS OF PER-O-ACETYLATED POLYOLS ON DB-23

GC peak	Per-O-acetyl derivative of	Retention time (min)	Relative retention time ^a		Methylene units ^b
			RRT ¹	RRT ²	
1	1,5-Anhydroxylitol	33.07	0.811	0.667	24.75
2	1,5-Anhydroribitol	34.13	0.837	0.682	25.29
3	1,5-Anhydroarabinitol	34.39	0.844	0.687	25.42
4	1,4;3,6-Dianhydroglucitol	34.60	0.849	0.692	25.53
5	Erythritol + 1,4-anhydroarabinitol	34.87	0.856	0.697	25.63
6	1,4-Anhydroxylitol	35.06	0.860	0.701	25.74
7	1,4;3,6-Dianhydromannitol	35.24	0.864	0.705	25.85
8	1,4-Anhydroribitol	36.03	0.884	0.720	26.26
9	Threitol	36.29	0.890	0.726	26.43
10	1,5;3,6-Dianhydroglucitol	36.54	0.896	0.731	26.57
11	1,4-Anhydrolyxitol	36.68	0.900	0.733	26.66
12	Pentaerythritol	40.77	1.000	0.815	29.22
13	Ribitol	41.67	1.022	0.833	29.65
14	Arabinitol	42.02	1.031	0.840	29.76
15	1,4-Anhydroglucitol	42.32	1.038	0.846	29.86
16	1,5-Anhydroglucitol	42.91	1.052	0.858	30.29
17	1,4-Anhydrogalactitol	43.18	1.059	0.863	30.43
18	1,4-Anhydromannitol + 1,5-anhydrogalactitol	43.34	1.063	0.866	30.50
19	1,5-Anhydromannitol	43.49	1.067	0.869	30.57
20	Xylitol + 2,5-anhydroglucitol	43.84	1.076	0.876	30.74
21	1,6-Anhydroiditol	45.28	1.111	0.905	32.26
22	Allitol	46.35	1.137	0.927	32.58
23	Mannitol	47.16	1.157	0.943	32.84
24	Altritol	47.35	1.161	0.947	32.87
25	Galactitol	48.03	1.178	0.960	34.07
26	Glucitol	49.05	1.203	0.981	34.38
27	Myoinositol	50.02	1.227	1.000	34.68
28	Iditol	50.89	1.248	1.017	34.97

^a RRT¹ relative to pentaerythritol, RRT² relative to myoinositol.

^b Measured relative to C₂₄–C₃₅ n-alkanes from 80°C at 3°C/min.

1,5-anhydro-D-galactitol [47]; 1,4-anhydro-D-glucitol [47]; 1,5-anhydro-D-glucitol [47]; 2,5-anhydro-D-glucitol [47]; 1,6-anhydro-L-iditol [47]; 1,4-anhydro-D,L-xylitol [44]; 1,5-anhydro-xylitol [44]; 1,4-anhydro-D-lyxitol [44]; 1,4-anhydro-D-mannitol [47]; 1,5-anhydro-D-mannitol [47]; 1,4-anhydro-D,L-ribitol [44]; 1,5-anhydroribitol [44]; D-arabinitol (Sigma); 1,4,3,6-dianhydro-D-glucitol [47]; 1,5,3,6-dianhydro-D-glucitol [47]; 1,4,3,6-dianhydro-D-mannitol [47]; erythritol (Aldrich); galactitol (Sigma); D-glucitol (Sigma); L-iditol [48]; xylitol (Sigma); D-mannitol (Sigma); myoinositol (Sigma); pentaerythritol (Sigma); ribitol (Sigma); D,L-threitol (Aldrich); bis(trimethylsilyl)acetamide (BSTFA) (Aldrich); acetic anhydride (Polskie Odczynniki Chemiczne, Gliwice, Poland); sodium acetate (Polskie Odczynniki Chemiczne); C₁₀, C₁₂, C₁₄, C₁₆, C₁₈, C₂₀, C₂₂, C₂₄, C₂₈, C₃₂, C₃₆ *n*-alkanes kit (Alltech); C₉, C₁₁, C₁₃, C₁₅, C₁₇, C₁₉, C₂₁, C₂₃, C₂₅

n-alkanes kit (Alltech); *n*-hexacosane (Alltech); *n*-triacontane (Alltech); and tetratriacontane (Alltech).

Solvents

Chloroform, pyridine and hexane were obtained from Aldrich.

Apparatus and columns

A VEGA 6180 gas chromatograph was obtained from Fisons Instruments (Vienna, Austria), a DB-23 capillary column from J&W Scientific (Folsom, CA, USA) and an HP-5 capillary column from Hewlett-Packard (Geneva, Switzerland).

Preparation of standard mixture

About 2-5 mg of each of the 31 reactants were dissolved in 10 ml of water and 20 μ l of this mixture were placed in a Reacti-vial. Water was then removed under a stream of nitrogen and 0.5

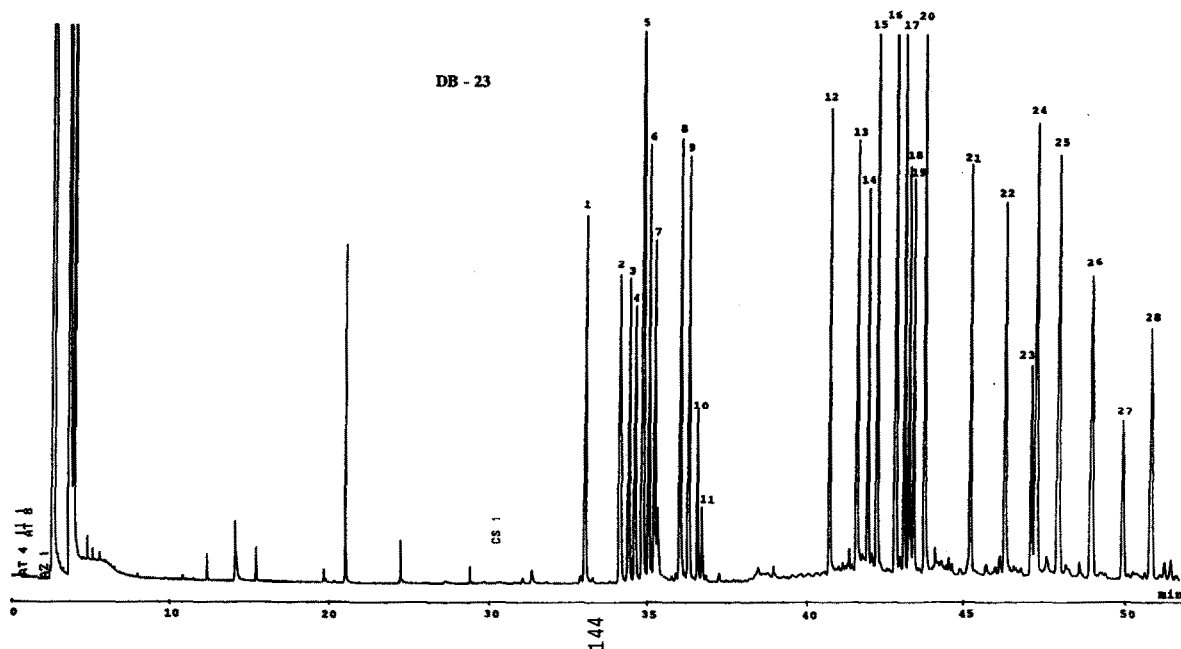


Fig. 1. Gas chromatogram of per-O-acetyl derivatives of the 31-component mixture separated on DB-23. Peaks: 1 = 1,5-anhydroxylitol; 2 = 1,5-anhydroribitol; 3 = 1,5-anhydroarabinitol; 4 = 1,4,3,6-dianhydroglucitol; 5 = erythritol + 1,4-anhydroarabinitol; 6 = 1,4-anhydroxylitol; 7 = 1,4,3,6-dianhydromannitol; 8 = 1,4-anhydroribitol; 9 = threitol; 10 = 1,5,3,6-dianhydroglucitol; 11 = 1,4-anhydrolyxitol; 12 = pentaerythritol; 13 = ribitol; 14 = arabinitol; 15 = 1,4-anhydroglucitol; 16 = 1,5-anhydroglucitol; 17 = 1,4-anhydrogalactitol; 18 = 1,4-anhydromannitol + 1,5-anhydrogalactitol; 19 = 1,5-anhydromannitol; 20 = xylitol + 2,5-anhydroglucitol; 21 = 1,6-anhydroiditol; 22 = allitol; 23 = mannitol; 24 = alritol; 25 = galactitol; 26 = glucitol; 27 = myoinositol; 28 = iditol.

ml of freshly distilled acetic anhydride and 10 mg of anhydrous sodium acetate were added. The mixture was heated at 100°C for 1 h, then the volatile components were removed under reduced pressure and the residue was dissolved in 100 μ l of chloroform.

Another 20 μ l portion of the same mixture, after removing water under a stream of nitrogen, was O-trimethylsilylated with 100 μ l of BSTFA in 100 μ l of pyridine at 100°C for 15 min.

RESULTS AND DISCUSSION

The separation with the optimum temperature programme and with the optimum carrier gas flow-rate of the per-O-acetylated 31-component mixture of polyol compounds on a fused-silica cGC column with the stationary phase DB-23 (Table I) gave 28 well separated peaks (Fig. 1). Erythritol and 1,4-anhydroarabinitol (peak 5), 1,4-anhydromannitol and 1,5-anhydrogalactitol

(peak 18) and xylitol and 2,5-anhydroglucitol (peak 20) were not separated. It is noteworthy that a separately prepared mixture of per-O-acetyl-1,4-anhydro-D-mannitol and -1,5-anhydro-D-galactitol could be separated on a glass column coated, according to Grob and Grob's procedure [49], with a mixed Carbowax 20 MTPA-SP 2340 (2:1, w/w) stationary phase. The remaining components were separated to the baseline and could be quantitatively determined.

The elution order of per-O-acetylated compounds (Fig. 1, Table I), in particular polyol groups, is as follows:

(a) alditols: erythritol (5), threitol (9), ribitol (13), arabinitol (14), xylitol (20), allitol (22), mannitol (23), altritol (24), galactitol (25), glucitol (26) and iditol (28);

(b) anhydropentitols: 1,5-anhydroxylitol (1), -ribitol (2) and -arabinitol (3), 1,4-anhydroarabinitol (5), -xylitol (6), -ribitol (8) and -lyxitol (9); the elution order of fully acetylated

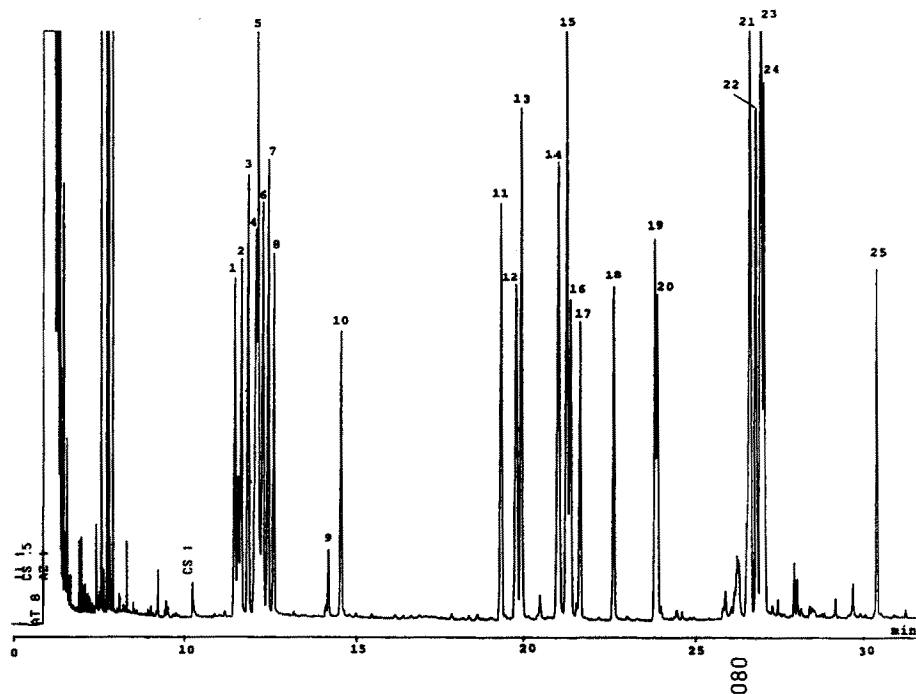


Fig. 2. Gas chromatogram of per-O-trimethylsilyl derivatives of the 31-component mixture separated on HP-5. Peaks: 1 = 1,5-anhydroarabinitol; 2 = 1,5;3,6-dianhydroglucitol + 1,4;3,6-dianhydroglucitol; 3 = 1,4-anhydroarabinitol + threitol; 4 = pentaerythritol + erythritol; 5 = 1,4;3,6-dianhydromannitol; 6 = 1,5-anhydroribitol; 7 = 1,4-anhydroribitol; 8 = 1,4-anhydroxylitol; 9 = 1,4-anhydrolyxitol; 10 = 1,5-anhydroxylitol; 11 = xylitol; 12 = arabinitol; 13 = ribitol; 14 = 2,5-anhydroglucitol + 1,4-anhydrogalactitol; 15 = 1,4-anhydroglucitol; 16 = 1,5-anhydromannitol; 17 = 1,6-anhydroiditol; 18 = 1,5-anhydrogalactitol; 19 = 1,4-anhydromannitol; 20 = 1,5-anhydroglucitol; 21 = mannitol; 22 = glucitol; 23 = allitol + galactitol + iditol; 24 = altritol; 25 = myoinositol.

1,4-anhydropentitols is in accordance with the values of their free energy differences [42];

(c) anhydroglucitols: 1,4- (peak 15), 1,5- (peak 16) and 2,5- (peak 20);

(d) anhydrogalactitols: 1,4- (peak 17) and 1,5- (peak 18);

(e) anhydromannitols: 1,4- (peak 18) and 1,5- (peak 19). Thus, among these isomeric per-O-acetyl-anhydropentitols, 1,5-anhydropentitols elute before 1,4-anhydropentitols. Further, per-O-acetyl-1,4-anhydrohexitols isomers elute before the 1,5-anhydro isomers.

The use of a fused-silica cGC column with HP-5 stationary phase for the separation of the per-O-trimethylsilylated mixture of the same polyols compounds gave only 25 peaks (Fig. 2, Table II). In this case five sets of compounds could not be separated: 1,4;3,6- and 1,5;3,6-dianhydroglucitol (peak 2), 1,4-anhydroarabinitol and threitol (peak 3), erythritol and pentaerythritol (peak 4), 2,5-anhydroglucitol and 1,4-anhydrogalactitol (peak 14) and allitol, galactitol and iditol (peak 23). Thus, only twelve peaks were separated to the base line (7–13, 18, 21, 22

TABLE II
RETENTION CHARACTERISTICS OF PER-O-TRIMETHYLSILYLATED POLYOLS ON HP-5

GC peak	Per-O-trimethylsilyl derivative of	Retention time (min)	Relative retention time ^a		Methylene units ^b
			RRT ¹	RRT ²	
1	1,5-Anhydroarabinitol	11.60	0.380	0.950	15.17
2	1,5;3,6-Dianhydroglucitol + 1,4;3,6-dianhydroglucitol	11.79	0.386	0.966	15.23
3	1,4-Anhydroarabinitol + threitol	11.98	0.392	0.981	15.29
4	Pentaerythritol + erythritol	12.21	0.400	1.000	15.35
5	1,4;3,6-Dianhydromannitol	12.27	0.402	1.005	15.37
6	1,5-Anhydroribitol	12.40	0.406	1.016	15.42
7	1,4-Anhydroribitol	12.55	0.411	1.028	15.46
8	1,4-Anhydroxylytol	12.71	0.416	1.041	15.51
9	1,4-Anhydroxyitol	14.27	0.467	1.169	16.00
10	1,5-Anhydroxylytol	14.66	0.480	1.201	16.11
11	Xylitol	19.43	0.636	1.591	17.48
12	Arabinitol	19.88	0.651	1.628	17.60
13	Ribitol	20.04	0.656	1.641	17.66
14	2,5-Anhydroglucitol + 1,4-anhydrogalactitol	21.09	0.691	1.727	17.97
15	1,4-Anhydroglucitol	21.36	0.699	1.749	18.00
16	1,5-Anhydromannitol	21.45	0.702	1.757	18.04
17	1,6-Anhydroiditol	21.76	0.713	1.782	18.13
18	1,5-Anhydrogalactitol	22.71	0.744	1.860	18.42
19	1,4-Anhydromannitol	23.91	0.783	1.958	18.78
20	1,5-Anhydroglucitol	23.99	0.786	1.965	18.81
21	Mannitol	26.73	0.875	2.189	19.75
22	Glucitol	26.92	0.881	2.205	19.82
23	Allitol + galactitol + iditol	27.07	0.886	2.217	19.88
24	Altritol	27.14	0.889	2.223	19.93
25	Myoinositol	30.54	1.000	2.501	21.29

^a RRT¹ relative to myoinositol, RRT² relative to pentaerythritol.

^b Measured relative to C₁₅–C₂₂ *n*-alkanes from 115°C at 3°C/min.

and 25) and only these could be quantitatively analysed.

The elution order of per-O-trimethylsilylated compounds (Fig. 2, Table II), in particular polyol groups, is as follows:

(a) alditols: threitol (3), erythritol (4), xylitol (11), arabinitol (12), ribitol (13), mannitol (21), glucitol (22) and altritol (24);

(b) anhydropentitols: 1,5-anhydroarabinitol (1), 1,4-anhydroarabinitol (3), 1,5-anhydro-ribitol (6), 1,4-anhydroribitol (7), 1,4-anhydro-xylitol (8), 1,4-anhydrolyxitol (9) and 1,5-anhydroxylitol (10);

(c) anhydroglucitols: 2,5- (peak 14), 1,4- (peak 15) and 1,5- (peak 20);

(d) anhydrogalactitols: 1,4- (peak 14) and 1,5- (peak 18);

(e) anhydromannitols: 1,5- (peak 16) and 1,4- (peak 19).

The identities of all peaks in both instances were determined by the standard cGC co-injection method.

The relative retention times with respect to pentaerythritol and myoinositol and the number of methylene units were determined in the usual manner for both types of derivatives and are given in Tables I and II.

On the basis of this and previous work, we can conclude that the cGC separations of the per-O-acetyl derivatives of polyols give better results than for their O-trimethylsilyl derivatives. Hence these derivatives can be used for the determination of these compounds in biological fluids [50].

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REFERENCES

- D.H. Lewis, *Occurrence and Distribution of Storage Carbohydrates in Vascular Plants*, Cambridge University Press, Cambridge, 1984, p. 1.
- F. Keller, *J. Plant Physiol.*, 134 (1989) 141.
- N. Shibuya, *J. Chromatogr.*, 208 (1981) 96.
- E.V. Evtushenko and Yu.S. Ovadov, *Khim. Prir. Soedin.*, 1 (1982) 23.
- P.A.J. Gorin, A.J. Griblin, G.P. Slater and L. Hogge, *Carbohydr. Res.*, 106 (1982) 235.
- A. Okahira and H. Kabatake, *Bunseki Kagaku*, 31 (1982) 304.
- J. Klok, H.C. Cox, J.W. de Leeuw and P.A. Shenck, *J. Chromatogr.*, 253 (1982) 55.
- J.A. Lomax and J. Conchie, *J. Chromatogr.*, 236 (1982) 385.
- R. Geyer, H. Geyer, S. Kuehnhardt, W. Mink and S. Stirn, *Anal. Biochem.*, 121 (1982) 263.
- A. Okahira and H. Kobatake, *Bunseki Kagaku*, 32 (1983) T-83.
- G.M. Iskander and O.H. Ibn, *Dokl. Bolg. Akad. Nauk*, 36 (1983) 791.
- A. Bacic, P.J. Harris, E.W. Hak and A.E. Clarke, *J. Chromatogr.*, 315 (1984) 373.
- P.J. Harris, A. Bacic and A.E. Clarke, *J. Chromatogr.*, 350 (1985) 304.
- J.R. Hudson, S.L. Morgan and A. Fox, *J. High Resolut. Chromatogr. Chromatogr. Commun.*, 5 (1982) 285.
- J. Lehrfeld, *Anal. Chem.*, 56 (1984) 1803.
- T. Kiho, S. Ukai and Ch. Hara, *J. Chromatogr.*, 369 (1986) 415.
- H.J. Chaves das Neves, J. Hiquinaldo and A. Frank, *Rev. Port. Quim.*, 27 (1985) 474.
- P. Englmaier, *Carbohydr. Res.*, 144 (1985) 177.
- W.R. Sherman, K.E. Ackerman, R.A. Berger, B.G. Gish and M. Zinbo, *Biomed. Environ. Mass Spectrom.*, 13 (1986) 333.
- N.F. Laker, *J. Chromatogr.*, 184 (1980) 457.
- B.R. Pettit, G.S. King and K. Blau, *Biomed. Mass Spectrom.*, 7 (1980) 309.
- J. Kuśmiercz, J.J. DeGeorge, D. Sweeney, C. May and S.I. Rapoport, *J. Chromatogr.*, 497 (1989) 39.
- J. Roboz, D. Kappatos, C. Diana, J. Greaves and J.F. Holland, *Clin. Chem. (Winston-Salem, N.C.)*, 30 (1984) 1611.
- T. Niwa, N. Yamamoto, K. Maeda, K. Yamada and T. Ohki, *J. Chromatogr.*, 277 (1983) 25.
- T. Niwa, K. Yamada, T. Ohki, A. Saito and M. Mori, *J. Chromatogr.*, 336 (1984) 345.
- M. Tomana, J. Prchal, L.C. Garner, H.W. Skalka and S.A. Baker, *J. Lab. Clin. Med.*, 103 (1984) 137.
- H. Ono, M. Funahashi and S. Hayano, *Nippon Ganka Gakkai Zasshi*, 85 (1981) 1475.
- S. Yoshioka, S. Saitoh, Ch. Negishi, T. Fujisawa and A. Fujimari, *Boei Ika Daigakko Zasshi*, 7 (1982) 185.
- A. Olano, M.M. Calvo and G. Reglero, *Chromatographia*, 21 (1986) 538.
- R.G. Binder and W.F. Haddon, *Carbohydr. Res.*, 129 (1984) 21.
- C.W. Ford, *J. Chromatogr.*, 333 (1985) 167.
- H.N. Englyst and J.H. Cummings, *J. Assoc. Off. Anal. Chem.*, 71 (1988) 808.
- Y. Fukazawa and M. Iwashita, *Yamanashi-Kenritsu Eisei Kogay Kenkyusho Nenpo*, 28 (1984) 1.
- R.J. Henry, P.J. Harris, A.B. Blakeney and B.A. Stone, *J. Chromatogr.*, 262 (1983) 249.

- 35 N. Oi, T. Doi, H. Kitahara and Y. Inada, *J. Chromatogr.*, 208 (1981) 404.
- 36 A.L. Leavitt and W.R. Sherman, *Carbohydr. Res.*, 103 (1982) 203.
- 37 A. Leavitt, W.R. Sherman and R. William, *Methods Enzymol.*, 89 (1982) 3.
- 38 A. Koenig and I. Benecke, *J. Chromatogr.*, 269 (1983) 19.
- 39 J. Roboz, E. Nieves and J.F. Holland, *J. Chromatogr.*, 500 (1990) 413.
- 40 J. Szafranek and A. Wiśniewski, *J. Chromatogr.*, 187 (1980) 131.
- 41 A. Wiśniewski, J. Szafranek and J. Sokołowski, *Carbohydr. Res.*, 97 (1981) 229.
- 42 A. Wiśniewski, J. Gajdus, J. Sokołowski and J. Szafranek, *Carbohydr. Res.*, 114 (1983) 11.
- 43 A. Wiśniewski, J. Sokołowski and J. Szafranek, *J. Carbohydr. Chem.*, 2 (1983) 293.
- 44 A. Wiśniewski, E. Skorupowa, J. Sokołowski, D. Głód and G. Descotes, *J. Carbohydr. Chem.*, 8 (1989) 59.
- 45 J. Szafranek and A. Wiśniewski, *J. Chromatogr.*, 161 (1978) 213.
- 46 A. Wiśniewski, E. Skorupowa and J. Sokołowski, *J. Carbohydr. Chem.*, 10 (1991) 77.
- 47 A. Wiśniewski, E. Skorupowa, J. Sokołowski, L. Róžański, D. Głód and G. Descotes, *J. Carbohydr. Chem.*, 8 (1989) 73.
- 48 A.S. Meyer and T. Reichstein, *Helv. Chim. Acta*, 29 (1946) 152.
- 49 K. Grob and G. Grob, *J. Chromatogr.*, 125 (1976) 471.
- 50 A. Wiśniewski, J. Madaj, E. Skorupowa, J. Sokołowski, Z. Hejka, W. Dużyński and J. Sławek, *J. Chromatogr.*, in preparation.